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*Metallic Bonds and Metallic Properties 1. Is the following sentence true or false? Metals are made up of cations and valence electrons, not neutral atoms. 2. What are metallic bonds? 3. Name three properties of metals that can be explained by metallic bonding. a. b. c. 4. What happens to an ionic crystal when a force is applied to it? 5.*

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**SECTION 7.2 IONIC BONDS AND IONIC COMPOUNDS (pages 194–199)** *This section lists the characteristics of an ionic bond. It also describes the use of these characteristics to explain the electrical conductivity of ionic compounds when melted and when in aqueous solutions. Formation of Ionic Compounds (pages 194–195)* **1. What is an ionic bond? 2.**

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***Ionic and metallic bonds P.2/11 Formation of ionic bonds From atoms to ions Atoms can obtain the stable electronic arrangements of atoms of noble gases by gaining or losing electrons. A simple ion forms when an atom either loses or gains one or more electrons. It is either positively or negatively charged. [An ion can also be formed by two or more atoms joined together by covalent bonds first.***

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